Method to determine ash content of tall oil soap

Scope

This procedure is used to determine the residue after ignition at red heat.

Apparatus

- Crucible porcelain, silica or platinum, 70-mL capacity.
- 2. Desiccator.
- 3 Bunsen burner.
- 4. Muffle furnace.
- 5. Oven, general purpose, laboratory.

Reagents

Anhydrous ethanol or 3A ethanol (optional).

Procedure

1. Heat a clean crucible at 750° ± 25 °C in a muffle furnace for 20 minutes.

NOTE 1: A platinum crucible is preferred if the ash is to be analyzed later for metals.

- 2. Place the crucible in a desiccator and allow to cool to ambient temperature.
- 3. Weigh the crucible to the nearest 0.001 g. Record the Weight
 - 4. Place approximately $20~{\rm g}$ of the sample in the crucible and weigh to the nearest $0.001~{\rm g}$. Record this weight.
- 5. Pre-dry the sample overnight at 110°C in a drying oven.

- NOTE 2: If a more rapid determination is desired, do not pre-dry. Instead add 1-2 mL of anhydrous ethanol (3A ethanol may also be used). Then using a bunsen burner, gently heat the crucible until the sample is completely dry and carbonized if the sample ignites, allow to bum until the flame goes out.
- 6. Put the crucible and sample back into the muffle furnace for 30 minutes (at 750°C).
- 7. Return the crucible to the desiccator and allow to come to ambient temperature. Weigh the crucible and residue to the nearest 0.001 g. Record this weight.

Calculation

$$Ash, \% = \underbrace{(A - B)}_{C - B} \times 100$$

where:

A = Weight of crucible plus residue, g

B = Weight of crucible, g

C = Weight of crucible plus sample, g

Reference

ASTM D803 "Methods for Testing Tall Oil."